

**Short communication****A Simple Method of Determination of Partition Coefficient for Biologically Active Molecules**

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**Abstract.** A simple method is presented for the determination of partition coefficient of an effector between water environment and biological material, based on concentration-dependent effects. The method allows the determination of partition coefficients for biological objects such as algae, bacteria and other microorganisms.

**Key words:** Partition coefficient — Oxygen evolution rate — Algae — Trimecaine — *Chlorella vulgaris*

The understanding of a number of biological processes requires the knowledge of partition coefficients ( $K_p$ ) determining the distribution of an effector between biological material and water environment. Information on this partitioning can be obtained by calculating  $K_p$ .

Sedimentation or centrifugation of biological suspensions and measurements of the effector amounts in the supernatant are used for determination of  $K_p$ . To evaluate the effector concentration in the supernatant, various physico-chemical methods, including spectrophotometry, conductometry, polarography etc., can be used.

Indirect methods for the determination of  $K_p$ , requiring no phase separation procedures, can be divided in three groups.

i. The first group is based on studying changes of some physical or biochemical parameters of the biological material in dependence on effector concentration, using various amounts of biological material. Several authors have used this approach by following depression of phase transition temperature of phospholipid membranes (Kaminoh et al. 1988; Inoue et al. 1990; Gallová 1993) or changes of ultrasonic velocity in them (Babincová and Hianik 1994). Other authors estimated  $K_p$  by studying concentration-dependent properties of phospholipid membranes using molecular probes (Ondriaš et al. 1983; Lissi et al. 1989; Šeršeň et al. 1989). Another approach for the determination of the number of binding sites of an effector in chloroplasts was reported by Izawa and Good (1965) and Tischer and

Strotmann (1977) who exploited changes in  $IC_{50}$  values (i.e. effector concentrations in chloroplast suspensions inducing 50% inhibition of Hill reaction rate with respect to untreated controls) at various chloroplast concentrations.

*ii.* The second group of methods is based on the observation of spectral changes of an effector upon changing its concentration in the biological material (Welti et al. 1984; Balgavý et al. 1992).

*iii.* Also, the partition coefficient of an effector can be determined in the aqueous phase of biological suspensions, from the depression of its freezing temperature (Hill 1974; Vanderkooi et al. 1977).

In this report a simple method is presented for the determination of water/biological material  $K_p$ , using concentration-dependent inhibitory effect of trimecaine on oxygen evolution rate (OER) in the algal suspension of *Chlorella vulgaris*.

Trimecaine (2-diethylamino-2',4',6'-trimethylacetanilidinium chloride) was obtained from Slovafarma (Hlohovec, Slovakia), and was used without further purification.

The OER of algae was measured by a Clark type electrode (SOPS 31 atp., Chemoproject Prague) in a chamber prepared according to Bartoš et al. (1975) at 24°C. The composition of the applied algal medium is described in Sidóová et al. (1992). Irradiation (ca 100 W/m<sup>2</sup>) of the algal suspensions was carried out with a 250 W halogen lamp through a water filter. The algal suspension was accommodated in the dark for 4 h prior to OER measurements.

The relation between chlorophyll (Chl) content and the volume of algal cells was determined by centrifugation. The concentration of Chl (Chl<sub>a</sub> + Chl<sub>b</sub>) was estimated spectrophotometrically (SPECORD UV-VIS, Zeiss Jena, Germany) according to Inskeep and Bloom (1985) after extraction in N,N-dimethylformamide.

$K_p$  calculations are based on the assumption that effector concentrations in algal cells having the same inhibitory effects on OER must be equal, independent of the amount of algae in the suspension.

The formalism for the calculation of  $K_p$  starts from the definition

$$K_p = \frac{C_a}{C_w} = \frac{N_a \cdot V_w}{N_w \cdot V_a} \quad (1)$$

where  $C_i$  are concentrations, and  $N_i$  the numbers of the effector molecules in algal cells or water, and  $V_i$  are volumes of algae and water. The indices  $i$  denote the algal ( $a$ ) and aqueous ( $w$ ) phase, respectively. Taking into account that  $N_t = N_a + N_w$  ( $N_t$  is the total number of effector molecules in algal suspension) and using mathematical procedures, the  $N_a$  or ( $C_a$ ) can be expressed as follows

$$N_a = \frac{N_w V_a K_p}{V_w} \quad (2)$$

$$N_a = N_t \frac{V_a K_p}{V_w + V_a K_p} \quad (3)$$

$$C_a = C_t \frac{V_t K_p}{V_w + V_a K_p} \quad (4)$$

It is evident that the effector concentrations in algal cells must be equal (i.e.  $C_{a1} = C_{a2}$ ) at two different amounts of algae showing the same inhibitory activity of the effector (e.g. at  $IC_{50}$ ), and  $K_p$  may be calculated as follows

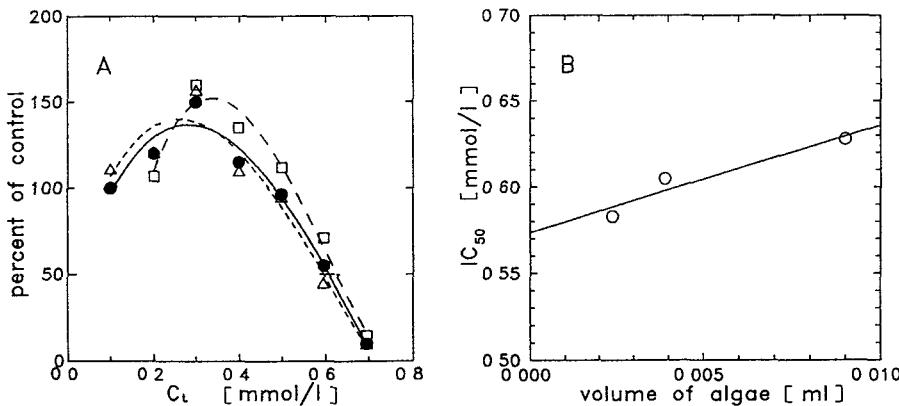
$$K_p = \frac{C_{t1} V_{t1} V_{w2} - C_{t2} V_{t2} V_{w1}}{C_{t2} V_{t2} V_{a1} - C_{t1} V_{t1} V_{a2}} \quad (5)$$

Provided that the total volumes  $V_t$  in all events are constant ( $V_t = 1$  ml) and  $V_w \gg V_a$  (this condition is met in our experiments because the algal concentrations employed are lower than 65 mg Chl/l, which corresponds to  $9.1 \times 10^{-3}$  ml of algal volume), whereby  $V_{w1} = V_{w2} = V_t$ ,  $K_p$  can be expressed as follows

$$K_p = \frac{(C_{t1} - C_{t2}) V_t}{C_{t2} V_{a1} - C_{t1} V_{a2}} \quad (6)$$

If  $V_{a2} = 0$  and  $C_{t2} = C_{t0}$ , eq. (6) can be modified to

$$K_p = \frac{(C_{t1} - C_{t0}) V_t}{C_{t0} V_{a1}} \quad (7)$$



**Figure 1.** A The dependences of OER in algal suspension upon trimecaine concentration at various amounts of algae. Squares, 64 mg Chl/l (long dashes); filled circles, 28 mg Chl/l (full line); and triangles, 17 mg Chl/l (short dashes). B The dependence of  $IC_{50}$  for trimecaine upon algal volume in suspension

After substitution of  $C_t$  by  $IC_{50}$  (the  $IC_{50}$  values were read from Fig. 1A at intersections of the horizontal dotted line with the experimental curves of the OER dependences upon effector concentrations), the  $K_p$  can be calculated by

$$K_p = \frac{V_t \cdot IC_{50}}{[IC_{50}]_0 \cdot V_{a1}} \quad (8)$$

where  $[IC_{50}]_0$  is the intercept with the ordinate, and  $IC_{50}/V_{a1}$  is the slope of the dependence of  $IC_{50}$  on algal volume (Fig. 1B).

Using centrifugation it was found that the algal suspension containing 1 mg Chl represents algal volume of 0.14 ml. By applying this Chl content-algal volume relation the partition coefficient for trimecaine  $K_p = 10.9 \pm 3.3$  was calculated, using the data presented in Fig. 1, according to formula (8). In parallel, the value of  $K_p = 9.8 \pm 2.5$  was determined by the centrifugation method using changes of trimecaine absorbance at 263 nm in the supernatant. It is obvious that the  $K_p$  values obtained by both methods are in a good accordance.

The present work shows that measurements of a certain quantitative property which can be immediately evaluated as the function of effector concentration (e.g. oxygen evolution or  $CO_2$  consumption in photosynthesis of algae,  $CO_2$  evolution or oxygen consumption in other microorganisms) can in general be used to determine  $K_p$  for living biological objects such as algae, bacteria or other microorganisms. If, for some reason, the phase separation method cannot be used, the above method of  $K_p$  determination can be applied.

## References

- Babincová M., Hianik T. (1994): Acoustical determination of a solute partition coefficient between two immiscible solvents. *Z. Phys. Chem.* **185**, 145—148
- Balgavý P., Benedikovič I., Kopecká B., Gallová J. (1992): Partition of piperidinoethylesters of 2-alkoxyphenylcarbamic acid in unilamellar phosphatidylcholine liposomes. *Gen. Physiol. Biophys.* **11**, 269—272
- Bartoš J., Berková E., Šetlík I. (1975): A versatile chamber for gas exchange measurements in suspensions of algae and chloroplasts. *Photosynthetica* **9**, 395—406
- Gallová J. (1993): The study of the influence of admixture molecules on model membranes using EPR spectroscopy and microcalorimetry. PhD Thesis. Faculty of Mathematics and Physics, Comenius University, Bratislava (in Slovak)
- Hill M. W. (1974): The effect of anaesthetic-like molecules on phase transition in smectic mesophases of dipalmitoyllecithin I. The normal alcohol up to C = 9 and three inhalation anaesthetic. *Biochim. Biophys. Acta* **356**, 117—124
- Inoue T., Fukushima K., Shimozawa R. (1990): Surfactant partition between bulk water and DPPC vesicle membrane: solid-gel vs. liquid-crystalline membrane. *Chem. Phys. Lipids* **52**, 157—161
- Inskeep W. P., Bloom P. R. (1985): Extinction coefficients of chlorophyll a and b in N,N-dimethyl formamide and 80% acetone. *Plant. Physiol.* **77**, 483—485

- Izawa, S., Good, N. E. (1965): The number of sites sensitive to 3-(3,4-dichlorophenyl)-1,5-dimethylurea, 3-(4-chlorophenyl)-1,1-dimethyl urea and 2-chloro-4-(2-propylamino)-6-ethyl-amino-s-triazine in isolated chloroplasts. *Biochim. Biophys. Acta* **102**, 20—38
- Kaminoh Y., Tashiro C., Kamaya H., Ueda I. (1988): Depression of phase-transition temperature by anesthetics: nonzero solid membrane binding. *Biochim. Biophys. Acta* **946**, 215—220
- Lissi E., Bianconi M. L., do Amaral A. T., de Paula E., Blanch L. E. B., Schreier S. (1989): Methods for the determination of partition coefficients based on the effect of solutes upon membrane structure. *Biochim. Biophys. Acta* **1021**, 46—50
- Ondriaš K., Balgavý P., Štolačka S., Horvath L. I. (1983): A spin label study of perturbation effect of tertiary amine anesthetics on brain lipid liposomes and synaptosomes. *Biochim. Biophys. Acta* **732**, 627—635
- Sidóová E., Králová K., Mitterhauszerová L. (1992): 3-Alkyl derivatives of 6-acetamido- and 6-nitro-2-benzothiazolinones as plant growth stimulators. *Chem. Papers* **46**, 55—58
- Šeršeň F., Leitmanová A., Devínsky F., Lacko I., Balgavý P. (1989): A spin label study of perturbation effects of N-(1-methyldodecyl)-N,N,N-trimethylammonium bromide and N-(1-methyldodecyl)-N,N-dimethylamine oxide on model membranes prepared from *Escherichia coli* — isolated lipids. *Gen. Physiol. Biophys.* **8**, 133—156
- Tischer W., Strotmann H. (1977): Relationship between inhibitor binding by chloroplasts and inhibition of photosynthetic electron transport. *Biochim. Biophys. Acta* **460**, 113—125
- Vanderkooi J. M., Landersberg R., Selick II H., McDonald G. G. (1977): Interaction of general anesthetics with phospholipid vesicles and biological membranes. *Biochim. Biophys. Acta* **464**, 1—16
- Welti R., Mullikin L. J., Yoshimura T., Helmckamp Jr. G. M. (1984): Partition of amphiphilic molecules into phospholipid vesicles and human erythrocyte ghosts: Measurements by ultraviolet difference spectroscopy. *Biochemistry USA* **23**, 6086—6091

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